# What makes Quantum Physics different from earlier Physics?

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**Quantum physics vs classical physics** 

# The experiments The theory Bonus track

**Quantum physics vs classical physics** 

# **1. The experiments**

# 2. The theory

# **3. Bonus track**

# An experiment with bullets



# An experiment with waves



# An experiment with electrons



# Let's watch the electrons



 $\Delta p \, \Delta x \ge h$ 

Summary of first principles:

(1) The probability of an event is  $P=|\psi|^2$ , where  $\psi$  is a complex number we call amplitude.

(2) If an event can take place in different ways, then  $P=|\sum_{i} \Psi_{i}|^{2}$ .

(3) If an experiment is able of determining which alternative is taken, then  $P = \sum_i P_i$ .

Why?

We only talk about predicting odds

Seems to explain everything on atomic scale

#### **Quantum physics vs classical physics**

## **1. The experiments**

# 2. The theory

# **3. Bonus track**

#### Schrödinger's Equation

1900-1926 – The years of confusion

 $E = h\nu$ 

$$E=h
u$$
  $p=rac{h}{\lambda}$  Planck's equation

 $\Delta p \Delta x \ge h$ 

Heisenberg's uncertainty

1927 – Postulate of the differential equation

 $\frac{\partial^2 \Psi}{\partial t^2} = v^2 \, \frac{\partial^2 \Psi}{\partial r^2}$  $d\vec{p}$ Newton's 2nd law  $ec{F}=$ Wave equation - waves -

De Broglie's hypothesis

$$\begin{split} \Psi(x,t) &= A \, e^{i(kx - \omega t)} & \longrightarrow & \Psi(x,t) = A \, e^{i(px - Et)/\hbar} \\ k &= \frac{2\pi}{\lambda} & \omega = 2\pi\nu & \text{let's make it} \\ p &= \hbar k \\ E &= \hbar\omega \\ (\hbar &= h/2\pi) \end{split}$$

#### **Schrödinger's Equation**

Newton's 2nd law 
$$\vec{F} = \frac{d\vec{p}}{dt}$$
  $\frac{\partial^2 \Psi}{\partial t^2} = v^2 \frac{\partial^2 \Psi}{\partial x^2}$  Wave equation -waves -  
 $\Psi(x,t) = A e^{i(px-Et)/\hbar} \longrightarrow \frac{E^2}{\hbar^2} \Psi = v^2 \frac{p^2}{\hbar^2} \Psi$   
Relativistic particles:  $\int v = c$   
 $E = pc$   $\checkmark$   
Slow particles:  $\int v < c$   
 $E = p^2/2m$   $\bigstar$   
Guess:  $\alpha \frac{\partial \Psi}{\partial t} = \beta \frac{\partial^2 \Psi}{\partial x^2} \longrightarrow i\hbar \frac{-i}{\hbar} E\Psi = \frac{-\hbar^2}{2m} \left(\frac{-p^2}{\hbar^2}\right) \Psi$ 

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#### **Schrödinger's Equation**

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x,t) \Psi$$

$$i\hbar\frac{\partial\Psi}{\partial t} = \left(-\frac{\hbar^2}{2m}\nabla^2 + V(\vec{r},t)\right) \Psi$$

#### Physical-mathematical considerations:

1)  $\psi(\mathbf{r},t)$  is the (complex) amplitude associated with the probability of finding a particle at ( $\mathbf{r},t$ ). Call it *wave function*:

$$P(\vec{r},t) = \Psi(\vec{r},t)^* \Psi(\vec{r},t)$$

2) Total probability of finding the particle confined in a volume must be unity.

$$\int \Psi(\vec{r},t)^* \, \Psi(\vec{r},t) dV = 1 \qquad \text{normalization condition}$$

Then,  $\psi(\mathbf{r},t)$  must be square integrable. For bound particles, if  $\mathbf{r} \to \infty$  then  $\psi(\mathbf{r},t) \to 0$ .

3)  $\psi$ (**r**,t) must be well behaved.

3.1) Finite. 3.2) Single-valued. 3.3) Continuous. 1<sup>st</sup> derivatives too.

4) Trivial solution, meaningless.

#### Let's play!

$$i\hbar\frac{\partial\Psi}{\partial t} = \left(-\frac{\hbar^2}{2m}\nabla^2 + V(\vec{r},t)\right) \Psi$$

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Particle in a 1D box



$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x) = E \qquad \Psi(x,t) = f(x) g(t)$$

$$\begin{bmatrix} -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x) \end{bmatrix} f(x) = Ef(x) \qquad \text{Time-independent}$$
Schrödinger Equation - stationary states -

$$-\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}f(x) = Ef(x) \longrightarrow f(x) = A\sin kx + B\cos kx$$
  
Boundary conditions 
$$\begin{cases} f(0) = 0 & \longrightarrow f(x) = A\sin kx \\ f(L) = 0 & \longrightarrow 0 = \sin kL & \longrightarrow k = \frac{\pi n}{L}, \ n = 1, 2, 3 \dots \end{cases}$$
$$f_n(x) = A\sin\frac{n\pi x}{L} \qquad E_n = \frac{\hbar^2 \pi^2}{2mL^2}n^2$$

#### It's a quantum world

$$n = 1, 2, 3...$$

Energy is quantised



Quantised energies explain discrete atomic spectra:



H Balmer series, 1885

$$E_n = \frac{\hbar^2 \pi^2}{2mL^2} n^2$$

#### **Operators**







Schrödinger Equation

 $\hat{E}\Psi = \hat{H}\Psi$ 

All operators are Hermitian

Remember: We only talk about predicting odds

Mean value of a physical magnitude: <

$$A\rangle = \frac{\int \Psi^* \hat{A} \Psi dV}{\int \Psi^* \Psi dV}$$
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#### **Variational principle**

Let  $\psi$  be the eigenfunction of the Hamiltonian, with eigenvalue  $E_{exact}$ . For any function  $\Phi$  with the same boundary conditions, and well-behaved, it holds:

$$\langle E \rangle_{approx} = \frac{\int \Phi^* \hat{H} \Phi dV}{\int \Phi^* \Phi dV} \ge \frac{\int \Psi^* \hat{H} \Psi dV}{\int \Psi^* \Psi dV} = \langle E \rangle_{exact}$$

It is customary to build a basis set formed by linearly independent functions:  $\{\varphi_1, \varphi_2, \varphi_3 \ldots\}$  whose scalar product is given by  $\int \varphi_i^* \varphi_j dV$ , define an aproximate solution as  $\Phi = \sum_i c_i \varphi_i$  and obtain the coefficients by minimizing  $\langle E \rangle_{approx}$ 

Optimizing the basis set to reduce the required Hilbert space dimension has been the front line of Quantum Chemistry research for over 40 years now.

#### **Quantum Physics and Mathematics:**









- (\*) Linear algebra
- (\*) Algebra of creation/annihilation operators
- (\*) Numerical methods
- (\*) Group theory (symmetry and permutation groups)

# What makes Quantum Physics different from earlier Physics?

(1) Dual character particle-wave

- (2) No longer «deterministic». We calculate probable values
- (3) Simultaneous knowledge of some magnitudes impossible
- (4) Schrödinger equation replaces Newton 2<sup>nd</sup> law & wave eq.
- (5) Physical magnitudes have quantised values

# **1. The experiments**

# 2. The theory

## **3. Bonus track**

Let  $\psi$  be the eigenfunction of the Hamiltonian, with eigenvalue  $E_{exact}$ . For any function  $\Phi$  with the same boundary conditions, and well-behaved, it holds:

$$\langle E \rangle_{approx} = \frac{\int \Phi^* \hat{H} \Phi dV}{\int \Phi^* \Phi dV} \ge E_{exact}$$

We choose a trial function  $\Phi_{\alpha}$  with a set of adjustable parameters  $\alpha = \{\alpha_1, \alpha_2, \alpha_3, ...\}$ , and look for those which minimize the energy.

An efficient way must be found to calculate the multi-dimensional integral above.

$$\begin{array}{c|c}
\hat{H}\Psi(\vec{r}) = E\Psi(\vec{r}) \\
E_L(\vec{r}) = \frac{\hat{H}\Psi(\vec{r})}{\Psi(\vec{r})} \\
\end{array} \rightarrow \langle E \rangle_{approx} = \frac{\int |\Phi|^2 E_L dV}{\int |\Phi|^2 dV} \approx \frac{\sum |\Phi|^2 E_L dV}{\sum |\Phi|^2 dV}$$

Stochastic methods: draw random values within V, average integrand, minimize energy or variance

#### **Variational Quantum Montecarlo**

$$\langle E \rangle_{approx} = \frac{\sum |\Phi|^2 E_L dV}{\sum |\Phi|^2 dV} \qquad E_L(\vec{r}) = \frac{\hat{H}\Phi(\vec{r})}{\Phi(\vec{r})}$$

Smart sampling critical for efficient calculation





**Metropolis** algorithm: priority to points with largest  $|\Phi|^2$  values. Good results with **much** fewer points.

Standard technique for many-electron systems, but fails for charges of opposite sign.



$$\hat{H} = \sum_{i=e,h} \frac{\vec{p}_i^2}{2m} + V(\vec{r}_e) + V(\vec{r}_h) - \frac{1}{\epsilon r_{eh}}$$
$$\Phi = \Phi_e(\vec{r}_e) \Phi_h(\vec{r}_h) e^{-ar_{eh}}$$