#### **What makes Quantum Physics different from earlier Physics?**

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**Quantum physics vs classical physics**

# **1. The experiments 2. The theory 3. Bonus track**

**Quantum physics vs classical physics**

### **1. The experiments**

### **2. The theory**

### **3. Bonus track**

### **An experiment with bullets**



### **An experiment with waves**



### **An experiment with electrons**



### **Let's watch the electrons**



 $\Delta p \, \Delta x \geq h$ 

Summary of first principles:

(1) The probability of an event is  $P=|\psi|^2$ , where  $\psi$  is a complex number we call amplitude.

(2) If an event can take place in different ways, then  $P=|\sum_i \psi_i|^2$ .

(3) If an experiment is able of determining which alternative is taken, then P= $\sum\limits_{\mathsf{i}}$  P i .

*Why?*

*We only talk about predicting odds*

8 *Seems to explain everything on atomic scale*

#### **Quantum physics vs classical physics**

### **1. The experiments**

## **2. The theory**

### **3. Bonus track**

#### **Schrödinger's Equation**

1900-1926 – The years of confusion

 $E = h\nu$ 

$E = h\nu$	$p = \frac{h}{\lambda}$
Planck's equation	$p = \frac{h}{\lambda}$

$$
\boxed{\Delta p \Delta x \geq h}
$$

Heisenberg's uncertainty

1927 – Postulate of the differential equation

*Wave equation Newton's 2nd law - waves - - particles*  let's make it *dual* $p=\hbar k$  $E=\hbar\omega$  $(\hbar = h/2\pi)$ 

De Broglie's hypothesis

#### **Schrödinger's Equation**

Newton's 2nd law  
\n- particles  
\n
$$
\vec{F} = \frac{d\vec{p}}{dt}
$$
\n
$$
\frac{\partial^2 \Psi}{\partial t^2} = v^2 \frac{\partial^2 \Psi}{\partial x^2}
$$
\nwave equation  
\n
$$
\Psi(x, t) = A e^{i(px - Et)/\hbar} \longrightarrow \frac{E^2}{\hbar^2} \Psi = v^2 \frac{p^2}{\hbar^2} \Psi
$$
\nRelative particle:  
\n
$$
\Sigma = p c
$$
\nSlow particles:  
\n
$$
\Sigma = p^2 / 2m
$$
\nSlow particles:  
\n
$$
\frac{v}{E} = p^2 / 2m
$$
\n
$$
\frac{\partial \Psi}{\partial t} = \beta \frac{\partial^2 \Psi}{\partial x^2} \longrightarrow i\hbar \frac{-i}{\hbar} E \Psi = \frac{-\hbar^2}{2m} \left(\frac{-p^2}{\hbar^2}\right) \Psi
$$

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#### **Schrödinger's Equation**

$$
i\hbar\frac{\partial\Psi}{\partial t}=-\frac{\hbar^2}{2m}\frac{\partial^2\Psi}{\partial x^2}+V(x,t)\Psi
$$

$$
i\hbar\frac{\partial\Psi}{\partial t}=\left(-\frac{\hbar^2}{2m}\nabla^2+V(\vec{r},t)\right)\,\Psi
$$

#### **Physical-mathematical** considerations:

1)  $\psi(\mathbf{r},t)$  is the (complex) amplitude associated with the probability of finding a particle at (**r**,t). Call it *wave function*:

$$
P(\vec{r},t) = \Psi(\vec{r},t)^* \Psi(\vec{r},t)
$$

2) Total probability of finding the particle confined in a volume must be unity.

$$
\int \Psi(\vec{r},t)^* \Psi(\vec{r},t) dV = 1
$$
 *normalization condition*

Then,  $\psi(\mathbf{r},t)$  must be square integrable. For bound particles, if  $\mathbf{r} \rightarrow \infty$  then  $\psi(\mathbf{r},t) \rightarrow 0$ .

3)  $\psi(\mathbf{r},t)$  must be well behaved.

3.1) Finite.  $3.2)$  Single-valued.  $3.3)$  Continuous. 1<sup>st</sup> derivatives too.

4) Trivial solution, meaningless.

#### **Let's play!**

$$
i\hbar \frac{\partial \Psi}{\partial t} = \left(-\frac{\hbar^2}{2m}\nabla^2 + V(\vec{r}, t)\right)\Psi
$$

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Particle in a 1D box



$$
i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x) = E \qquad \Psi(x, t) = f(x) g(t)
$$
  

$$
\left[ -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x) \right] f(x) = E f(x)
$$
Time-independent  
Schrödinger Equation  
- stationary states -

$$
-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} f(x) = Ef(x) \longrightarrow f(x) = A \sin kx + B \cos kx
$$
  
\nBoundary\n
$$
f(0) = 0 \longrightarrow f(x) = A \sin kx
$$
\nconditions\n
$$
f(L) = 0 \longrightarrow 0 = \sin kL \longrightarrow k = \frac{\pi n}{L}, \quad n = 1, 2, 3 \dots
$$
\n
$$
f_n(x) = A \sin \frac{n\pi x}{L} \quad E_n = \frac{\hbar^2 \pi^2}{2mL^2} n^2
$$

#### **It's a quantum world**

$$
n=1,\,2,\,3\ldots
$$

Energy is quantised



Quantised energies explain discrete atomic spectra:



*H Balmer series, 1885*

$$
E_n = \frac{\hbar^2 \pi^2}{2mL^2} n^2
$$

#### **Operators**







Schrödinger Equation  $\hat{E}\Psi = \hat{H}\Psi$ 

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All operators are Hermitian

Remember: *We only talk about predicting odds*

**Mean value** of a physical magnitude:  $\langle \ \rangle$ 

$$
A\rangle = \frac{\int \Psi^* \hat{A} \Psi dV}{\int \Psi^* \Psi dV}
$$

#### **Variational principle**

Let  $\psi$  be the eigenfunction of the Hamiltonian, with eigenvalue  $\mathsf{E}_{_{\sf exact}}$ . For any function Φ with the same boundary conditions, and well-behaved, it holds:

$$
\langle E\rangle_{approx}=\frac{\int\Phi^*\hat{H}\Phi dV}{\int\Phi^*\Phi dV}\geq \frac{\int\Psi^*\hat{H}\Psi dV}{\int\Psi^*\Psi dV}=\langle E\rangle_{exact}
$$

whose scalar product is given by  $\quad \int \, \varphi_i^* \varphi_j dV \quad$  , define an aproximate solution as  $|\Phi=\sum c_i\varphi_i|$  and obtain the coefficients by It is customary to build a basis set formed by linearly independent functions: minimizing  $\langle E \rangle_{approx}$ 

*Optimizing the basis set to reduce the required Hilbert space dimension has been the front line of Quantum Chemistry research for over 40 years now.*

#### **Quantum Physics and Mathematics:**









- ( \* ) Linear algebra
- ( \* ) Algebra of creation/annihilation operators
- ( \* ) Numerical methods
- ( \* ) Group theory (symmetry and permutation groups)

### **What makes Quantum Physics different from earlier Physics?**

(1) Dual character particle-wave

(2) No longer «deterministic». We calculate probable values

(3) Simultaneous knowledge of some magnitudes impossible

(4) Schrödinger equation replaces Newton  $2^{nd}$  law & wave eq.

(5) Physical magnitudes have quantised values

#### **Quantum physics vs classical physics**

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Let  $\psi$  be the eigenfunction of the Hamiltonian, with eigenvalue  $\mathsf{E}_{_{\sf exact}}$ . For any function Φ with the same boundary conditions, and well-behaved, it holds:

$$
\langle E\rangle_{approx}=\frac{\int \Phi^*\hat{H}\Phi dV}{\int \Phi^*\Phi dV}\geq E_{exact}
$$

We choose a trial function  $\Phi_{_\alpha}$  with a set of adjustable parameters  $\alpha = \{\alpha_{_1},\alpha_{_2},\alpha_{_3}...\}$ , and look for those which minimize the energy.

An efficient way must be found to calculate the multi-dimensional integral above.

$$
\hat{H}\Psi(\vec{r}) = E\Psi(\vec{r})
$$
\n
$$
E_L(\vec{r}) = \frac{\hat{H}\Psi(\vec{r})}{\Psi(\vec{r})}
$$
\n
$$
\langle E \rangle_{approx} = \frac{\int |\Phi|^2 E_L dV}{\int |\Phi|^2 dV} \approx \frac{\sum |\Phi|^2 E_L dV}{\sum |\Phi|^2 dV}
$$

Stochastic methods: draw random values within V, average integrand, minimize energy or variance

#### **Variational Quantum Montecarlo**

$$
\langle E \rangle_{approx} = \frac{\sum |\Phi|^2 E_L dV}{\sum |\Phi|^2 dV} \qquad E_L(\vec{r}) = \frac{\hat{H}\Phi(\vec{r})}{\Phi(\vec{r})}
$$

Smart sampling critical for *efficient* calculation





Random sampling (between -1 and 1) **Metropolis** algorithm: priority to points with largest |Φ| 2 values. Good results with **much** fewer points.

Standard technique for many-electron systems, **but fails for charges of opposite sign**.



$$
\hat{H} = \sum_{i=e,h} \frac{\vec{p}_i^2}{2m} + V(\vec{r}_e) + V(\vec{r}_h) - \frac{1}{\epsilon r_{eh}}
$$

$$
\Phi = \Phi_e(\vec{r}_e)\Phi_h(\vec{r}_h) e^{-ar_{eh}}
$$